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## Key indicators

Single-crystal X-ray study
$T=149 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.006 \AA$
$R$ factor $=0.044$
$w R$ factor $=0.097$
Data-to-parameter ratio $=11.4$

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.
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## Chloro( $\eta^{5}$-cyclopentadienyl) $\left\{\left[\eta^{5}: \eta^{1}-2\right.\right.$-( $p$-tolylsulfonamido)ethyl]cyclopentadienyl\}titanium

The title compound, $\left[\mathrm{TiCl}\left(\mathrm{C}_{5} \mathrm{H}_{5}\right)\left(\mathrm{C}_{14} \mathrm{H}_{15} \mathrm{NO}_{2}\right)\right]$, has a fourcoordinate Ti atom tetrahedrally bonded to the N atom of the sulfonamide moiety, which is tethered by a substituted twocarbon chain to a bound cyclopentadiene ring, another cyclopentadiene ligand and one Cl atom. $\mathrm{The} \mathrm{Ti}-\mathrm{N}, \mathrm{Cl}$ bond lengths are 2.123 (3) and 2.3250 (13) $\AA$, respectively.

## Comment

The title compound, (I), is related to previously reported structures (Lensink, Gainsford \& Baxter, 2001; Lensink, Gainsford \& Brandsma, 2001; Lensink, 1998). The crystal structure of (I) consists of independent molecules (Fig. 1), with only a few weak intermolecular contacts, e.g. C17$\mathrm{H} 17 \cdots \mathrm{O} 1(x-1 / 2,1 / 2-y, z-1 / 2)$, with a $\mathrm{C} 17 \cdots \mathrm{O} 1$ distance of 3.422 (5) A.

(I)

The Ti atom has tetrahedral coordination through the N atom of the sulfonamide moiety, a cyclopentadiene (assumed to occupy one coordination site) linked via an ethyl linkage to the nitrogen, one chloride and another cyclopentadiene. The $\mathrm{Ti}-C_{g}$ distances ( $C_{g}$ being the cylopentadienyl centres) are 2.061 (2) and 2.063 (2) $\AA$, within the narrow range 2.01-2.08 $\AA$ observed previously (Lensink, Gainsford \& Baxter, 2001; Lensink, Gainsford \& Brandsma, 2001; Allen \& Kennard, 1993; ConQuest, 2001), with a $C_{g}-\mathrm{Ti}-\mathrm{C}_{g}{ }^{\prime}$ angle of 131.37 (9) ${ }^{\circ}$. The $\mathrm{Ti}-\mathrm{N}$ distance of 2.123 (3) $\AA$ is shorter than in two related compounds: $2.437 \AA$ in ( $\eta^{5}$-cyclopentadienyl) $\left\{\eta^{5}\right.$-[(2-N,N-dimethylamino)ethyl $]$ tetramethylcyclopentadienyl]chlorotitanium (CSD refcode NOGYAJ; Beckhaus et al., 1997) and $2.428 \AA$ in chloro( $\eta^{5}$-diaminoethylcyclopentadienyl-$\mathrm{N})\left(\eta^{5}\right.$-dimethylaminoethyl)cyclopentadienyl)titanium(III) (REXGOQ; Qian et al., 1997), but longer than the TiN (imino) distance found in ( $\eta^{5}$-2-isopropyliminoethylcyclopentadienyl)dichlorotitanium (NOMSUD; Sinnema et al., 1997). The $\mathrm{Ti}-\mathrm{Cl}$ distance is within the range found for other titanium complexes with bound cyclopentadienes (2.29$2.50 \AA$; Conquest, 2001): 2.325 (1) $\AA$ compared with 2.461 , 2.483 and $2.275 / 2.300 \AA$ in NOGYAJ, REXGOQ and NOMSUD, respectively. The lengthening of the $\mathrm{Ti}-\mathrm{N}$ bonds in these four compounds parallels the change from trigonal

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planar to trigonal pyramidal geometry of their N atoms [here N 1 is 0.277 (4) $\AA$ from the plane of Ti1, S1 and C8]. A more dominant factor for the change compared to that in NOMSUD is the increased intramolecular crowding: between the sulfonamide O atoms and cyclopentadienyl ring here, involving the two methyl groups bound to the nitrogen in NOGYAG and REXGOQ, and also the methyl groups on the cyclopentadienyl in REXGOQ.

## Experimental

The title compound, (I), was prepared from the stoichiometric reaction of $\left(\mathrm{C}_{14} \mathrm{H}_{15} \mathrm{NO}_{2} \mathrm{~S}\right) \mathrm{TiCl}_{2}$ with $\mathrm{NaC}_{5} \mathrm{H}_{5}$. Crystals were obtained from dichloromethane/pentane. Details and spectroscopic data will be reported in a future publication.

## Crystal data

$\left[\mathrm{TiCl}\left(\mathrm{C}_{5} \mathrm{H}_{5}\right)\left(\mathrm{C}_{14} \mathrm{H}_{15} \mathrm{NO}_{2} \mathrm{~S}\right)\right]$
$M_{r}=409.77$
Monoclinic, $P 2_{1} / n$
$a=7.9620$ (18) A
$b=18.734$ (2) $\AA$
$c=11.6482(15) \AA$
$\beta=98.678(11)^{\circ}$
$V=1717.5(5) \AA^{3}$
$Z=4$

$$
\begin{aligned}
& D_{x}=1.585 \mathrm{Mg} \mathrm{~m}^{-3} \\
& \text { Mo } K \alpha \text { radiation } \\
& \text { Cell parameters from } 8192 \\
& \quad \text { reflections } \\
& \theta=2.9-24.4^{\circ} \\
& \mu=0.79 \mathrm{~mm}^{-1} \\
& T=149(2) \mathrm{K} \\
& \text { Needle, yellow } \\
& 0.40 \times 0.10 \times 0.04 \mathrm{~mm}
\end{aligned}
$$

## Data collection

Siemens CCD area-detector diffractometer
$\varphi$ and $\omega$ scans
Absorption correction: multi-scan
(Sheldrick, 1996; Blessing, 1995)
$T_{\text {min }}=0.743, T_{\text {max }}=0.969$
6268 measured reflections

> 2551 independent reflections
> 1746 reflections with $I>2 \sigma(I)$
> $R_{\text {int }}=0.062$
> $\theta_{\max }=26.3^{\circ}$
> $h=-9 \rightarrow 9$
> $k=-23 \rightarrow 23$
> $l=-12 \rightarrow 10$

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.044$
$w R\left(F^{2}\right)=0.097$
$S=0.97$
2551 reflections
223 parameters

H-atom parameters constrained
$w=1 /\left[\sigma^{2}\left(F_{o}^{2}\right)+(0.039 P)^{2}\right]$
where $P=\left(F_{o}{ }^{2}+2 F_{c}{ }^{2}\right) / 3$
$(\Delta / \sigma)_{\max }<0.001$
$\Delta \rho_{\max }=0.29 \mathrm{e}_{\mathrm{\circ}} \AA^{-3}$
$\Delta \rho_{\text {min }}=-0.31 \mathrm{e}^{-3}$

Table 1
Selected geometric parameters $\left(\AA{ }^{\circ},{ }^{\circ}\right)$.

| Ti1-N1 | $2.123(3)$ | S1-N1 | $1.602(4)$ |
| :--- | :---: | :--- | :--- |
| Ti1-Cl1 | $2.3250(13)$ | $\mathrm{S} 1-\mathrm{C} 1$ | $1.779(4)$ |
| S1-O1 | $1.441(3)$ | $\mathrm{N} 1-\mathrm{C} 8$ | $1.491(5)$ |
| S1-O2 | $1.444(3)$ |  |  |
| N1-Ti1-Cl1 | $99.91(9)$ | $\mathrm{C} 8-\mathrm{N} 1-\mathrm{Ti} 1$ | $114.2(3)$ |
| $\mathrm{C} 8-\mathrm{N} 1-\mathrm{S} 1$ | $112.7(2)$ | $\mathrm{S} 1-\mathrm{N} 1-\mathrm{Ti} 1$ | $125.25(19)$ |

The completeness of data collection was relatively low, 0.730 for $\theta_{\max }=26.3^{\circ}$. Excessive build-up of ice, coupled with subsequent loss of the crystal, restricted the collection to the initial $360^{\circ}$ scan, leaving


Figure 1
The molecular structure of (I). Displacement ellipsoids are drawn at the 50\% probability level.
part of the reciprocal space not sampled. Sufficient redundancy of data permitted adequate absorption corrections. All H atoms, except those on methyl C atoms, were constrained to $U_{\text {iso }} 1.2$ times $U_{\text {eq }}$ of their parent atom. The factor was 1.5 times for the methyl H atoms.

Data collection: SMART (Siemens, 1996); cell refinement: SMART; data reduction: SAINT (Siemens, 1996) and SADABS (Sheldrick, 1996); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP-3 in WinGX (Farrugia, 1997, 1999); software used to prepare material for publication: SHELXL97.

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